

Crystallographic report

Bis{4-[N,N-bis(2-cyanoethyl)amino]pyridine} dichlorozinc(II)

Jun Ni, Wei Xu, Zheng Xue, Yi-Zhi Li, Hui-Lan Chen and Zhi-Lin Wang*

Coordination Chemistry Institute, State Key Laboratory of Coordination Chemistry, Nanjing University, Nanjing 210093, People's Republic of China

Received 8 February 2004; Revised 14 March 2004; Accepted 15 March 2004

In $(C_{11}H_{12}N_4)_2ZnCl_2$, the zinc(II) center is coordinated by the pyridine nitrogen atoms of two 4-[N,N-bis(2-cyanoethyl)amino]pyridine ligands and two chlorine atoms in a tetrahedral geometry. Copyright © 2004 John Wiley & Sons, Ltd.

KEYWORDS: 4-[N,N-bis(2-cyanoethyl)amino]pyridine; crystal structure; hydrogen bonds; zinc(II)

COMMENT

The 4-[N,N-bis(2-cyanoethyl)amino]pyridine molecule comprises a pyridine ring 'head' and two cyanoethyl group 'tails', and as such can be used to form different hydrogen bonding patterns leading to distinct packing structures.^{1–3} Here, we report a new mononuclear zinc complex, $(C_{11}H_{12}N_4)_2ZnCl_2$, as shown in Fig. 1. The zinc(II) atom is tetrahedrally coordinated by two pyridine nitrogen atoms and two chlorine atoms with the range of tetrahedral angles being 107.08(6) to 117.40(3)°. The crystal structure is stabilized by a complex network of $CN \cdots H$ interactions.

EXPERIMENTAL

A mixture of 4-[N,N-bis(2-cyanoethyl)amino]pyridine (400 mg, 2 mmol) and $ZnCl_2$ (136 mg, 1 mmol) was dissolved in methanol (25 ml). The reaction mixture was refluxed for 3 h and then filtered, yielding a white product. Yield: 456 mg (85%). Analysis. Found: C, 49.18; H, 4.47; N, 20.92. Calc.: C, 49.23; H, 4.51; N, 20.88%. The resultant colorless filtrate was left to stand at room temperature. Well-defined block-shaped colorless single crystals of the title compound were obtained within 1 week by slow evaporation of the above filtrate. Intensity data for $(C_{11}H_{12}N_4)_2ZnCl_2$ were collected at 293 K on a Bruker Smart Apex CCD diffractometer for a colorless crystal with dimensions of $0.10 \times 0.15 \times 0.20$ mm³. $C_{22}H_{24}Cl_2N_8Zn$, $M = 536.76$,

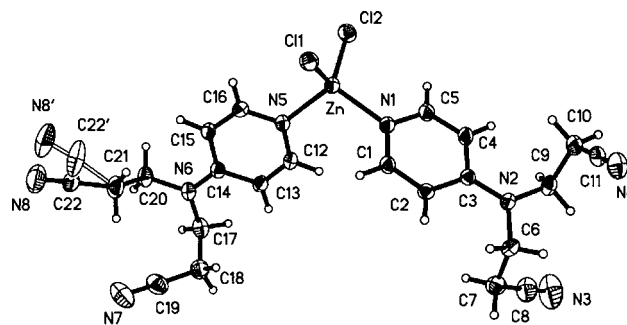


Figure 1. Molecule structure of $(C_{11}H_{12}N_4)_2ZnCl_2$, with displacement ellipsoids drawn at the 30% probability level. Selected bond distances (Å) and angles (°): Zn–Cl1 2.2411(8), Zn–Cl2 2.2429(8), Zn–N1 2.010(2), Zn–N5 2.033(2), Cl1–Zn–Cl2 117.40(3), Cl1–Zn–N1 108.50(7), Cl1–Zn–N5 107.24(7), Cl2–Zn–N1 108.17(6), Cl2–Zn–N5 107.08(6), N1–Zn–N5 108.12(9).

triclinic, $P\bar{1}$, $a = 9.447(1)$, $b = 10.699(1)$, $c = 12.231(1)$ Å, $\alpha = 94.79(1)$, $\beta = 90.10(1)$, $\gamma = 96.77(1)^\circ$, $V = 1223.2(2)$ Å³, $Z = 2$, 4244 unique data ($\theta_{\max} 25.0^\circ$), 3434 data with $I \geq 2\sigma(I)$, $R = 0.038$ (obs. data), and $wR = 0.093$ (all data). Programs used: SMART, SAINT, and SHELXTL. CCDC deposition number: 230571. The C22 and N8 atoms in one ligand are distorted and were refined over two sites with site occupancies, from refinement, equal to 0.567(16) and 0.433(16) respectively.

Acknowledgements

The National Science Foundation of China (G20171021) and the Specialized Research Fund for the Doctoral Program of Higher Education (G2000028436) are thanked for support.

*Correspondence to: Zhi-Lin Wang, Coordination Chemistry Institute, State Key Laboratory of Coordination Chemistry, Nanjing University, Nanjing 210093, People's Republic of China. E-mail: wangzl@nju.edu.cn

Contract/grant sponsor: National Science Foundation of China; Contract/grant number: G20171021.

Contract/grant sponsor: Specialized Research Fund for the Doctoral Program of Higher Education; Contract/grant number: G2000028436.

REFERENCES

1. Ni J, Li YZ, Qi WB, Liu YJ, Chen HL, Wang ZL. *Acta Crystallogr. Sect. C* 2003; **59**: o470.
2. Ni J, Li YZ, Xue Z, Chen HL, Wang ZL. *Acta Crystallogr. Sect. C* 2003; **59**: m201.
3. Geng JL, Ni J, Liu R, Chen HL, Wang ZL. *Acta Crystallogr. Sect. E* 2003; **59**: o1697.